# NEW MELAMPOLIDES AND DARUTIGENOL FROM SIGESBECKIA ORIENTALIS

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Abstract—Isolation and identification of darutigenol and two new melampolides from Sigesbeckia orientalis, in addition to the previously described orientalide and darutoside, are reported.

#### INTRODUCTION

Isolation of the melampolide orientalide (1a) from the medicinal plant Sigesbeckia orientalis L. has been reported earlier [1]. We now describe isolation of two new melampolides 1b and 4a and the diterpene darutigenol (8a) from the more polar fractions of the extracts, which also contained the known darutoside [3].

# RESULTS AND DISCUSSION

The non-crystalline lactone **1b**,  $C_{21}H_{24}O_7$  (high resolution MS) had a <sup>1</sup>H NMR spectrum (see Experimental) essentially superimposable on that of acanthospermal A (**1c**) [4] except for the signals of the two ester side chains which were acetate and methacrylate instead of isobutyrate and  $\alpha$ -hydroxyisobutyrate. This was corroborated by the MS (see Experimental). Se-

quences C-1 through C-3 and C-5 through C-9 were established by spin decoupling in the manner detailed previously [1, 4] as was the fact that the aldehyde function was conjugated with the 1,10-double bond, the latter being cis (chemical shift of H-14). Identity of the various coupling constants with those of the acanthospermals and orientalide indicated the stereochemistry shown in the formula. Allocation of the methacrylate to C-8 and the acetate to C-9 was based on analogy to 1c and 4a since selective hydrolysis of one of the ester groups could not be affected. Instead, treatment with KOH-MeOH afforded 2a by solvolytic displacement of the ester on C-9 in the manner previously observed with the acanthospermals and 1a, concomitant with hydrolysis of the ester on C-8 and addition of methanol to the conjugated lactone. In 2a attachment of the ring methoxy group to C-9 was verified by acetylation to 2b and oxidation (DMSO-Ac<sub>2</sub>O) to 3. In the <sup>1</sup>H NMR spectrum of 3,

1a R = OH, R' = 
$$\begin{pmatrix} 0 \\ 10 \\ 9 \\ 10 \\ 10 \end{pmatrix}$$
, R" = Ac

2a R,R' = H

1b R = H, R' =  $\begin{pmatrix} 0 \\ 11 \\ 0 \\ 0 \end{pmatrix}$ , R" = Ac

2b R = H, R' = Ac

1c R = H, R' =  $\begin{pmatrix} 0 \\ 11 \\ 0 \\ 0 \end{pmatrix}$ , R" = Ac

2d R = OH, R' = H

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the signal of the proton under the methoxyl group was a sharp singlet at 3.80 ppm and the signal at 7.01 ppm of the proton  $\beta$  to the aldehyde remained coupled to the protons of a methylene group found at 2.72 and 2.49 ppm. Consequently we can dismiss the possibility considered in our orientalide paper [1] that in this series solvolysis of the group attached to C-9 may be accompanied by allylic rearrangement. The product from methanolysis of 1a is therefore 2c, not 5.

The second lactone 4a, mp 208°,  $C_{20}H_{24}O_7$ , was an analog of orientalide containing a methoxyl group on C-9 (upfield shift of H-9 from 5.3 to 3.8 ppm). Acetylation gave 4b; hydrolysis with KOH-MeOH yielded 2c identical with material previously [1] obtained by methanolysis of 1a. MnO<sub>2</sub> oxidation of 2c gave the known 6a [1] which was further oxidized (DMSO-Ac<sub>2</sub>O) to 6b. The <sup>1</sup>H NMR spectrum of the latter confirmed the location of the methoxyl on C-9 and hence the absence of a rearrangement in the methanolysis of 1a. Exposure of 4a to  $BF_3$ -Ac<sub>2</sub>O gave a triacetate which is formulated as the enol acetate 7.

Darutigenol (8a), which has not been isolated previously from S. orientalis [2, 3], was identified through

conversion to the triacetate (8b), tribenzoate (8c) and the aldehyde 9. Its stereochemistry at C-15 has been established recently [5].

## **EXPERIMENTAL**

The extraction of S. orientalis has been described [1]. Since fractions 41-90 of the original chromatogram showed several spots on TLC, they were combined (10.9 g) and rechromatographed over 400 g Si gel, fractions being collected as follows: 1-10 (C<sub>6</sub>H<sub>6</sub>-EtOAc, 1:4), 11-20 (EtOAc), 21-30 (EtOAc-MeOH, 19:1) and 31-40 (EtOAc-MeOH, 9:1) Fractions 4-12 (2.18 g) contained three substances which were separated by PLC (C<sub>6</sub>H<sub>6</sub>-EtOAc, 1:2). The least polar substance 1b was a gum (0.55 g) which was not completely pure (vide infra) and had IR bands (CHCl<sub>3</sub>) at 2700 (-CHO), 1775 (lactone), 1730 (esters), 1690 ( $\alpha$ ,  $\beta$ -unsaturated aldehyde), 1650 and 1140 cm<sup>-1</sup>: UV strong end absorption  $(\varepsilon_{230} \ 18\ 600)$ . <sup>1</sup>H NMR (270 MHz CDCl<sub>3</sub>):  $\delta$  9.48 ( $J = 2 \ Hz$ , H-14), 6.75 (m, H-1), 6.75 (dd, J = 9, 1.5 Hz, H-8), 6.29 and 5.85(d, J = 3 Hz, H-13), 6.04 and 5.60 (br, H-3'), 5.33 (dd,J = 9, 2 Hz, H-9), 5.10 (t, J = 10 Hz, H-6), 4.92 (br, J =10 Hz, H-5), 2.84 and 2.48 (m, H-2), 2.65 (m, H-7), 2.04 (br,

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H-15), 1.94 (Ac) and 1.92 (br, H-4'). The signals of H-3 were submerged near 1.95 ppm. Additional weak signals indicated the presence of an impurity which could not be removed by PLC. The low resolution MS exhibited significant peaks at m/e 388 (M<sup>+</sup>), 328 (M-HOAc), 31 (M-C<sub>4</sub>H<sub>5</sub>O), 277 (M<sup>+</sup>-C<sub>4</sub>H<sub>5</sub>O-C<sub>2</sub>H<sub>2</sub>O), 259 (M<sup>+</sup>-C<sub>4</sub>H<sub>5</sub>O-HOAc), 242 (M<sup>+</sup>-C<sub>4</sub>H<sub>6</sub>O<sub>2</sub>-HOAc), 213 and 69 (C<sub>4</sub>H<sub>5</sub>O). (Calc. for C<sub>21</sub>H<sub>24</sub>O<sub>7</sub>: MW, 388.1520. Found: MW(MS), 388.1516).

The next substance **4a** was recrystallized from MeOH, mp 208° (0.42 g). IR bands at 2700, 1760, 1720, 1690 and 1180 cm<sup>-1</sup>, <sup>1</sup>H NMR (60 MHz):  $\delta$  9.60 (d, J=2 Hz, H-14), 6.82 (ddbr, J=9, 8 Hz, H-1), 6.65 (dd, J=9, 1 Hz, H-8), 6.28 (d, J=3 Hz, H-13 a), 6.10 (br, H-3'), 5.85 (d, J=3 Hz, H-13 b), 5.65 (br, H-3'), 5.10 (m, H-5 and H-6), 4.50 (br, H-15), 3.80 (dd, J=9, 2 Hz, H-9), 3.18 (OMe), 1.98 (br, H-4'). The MS exhibited significant peaks at m/e 376 (M\*), 358 (M\*-H<sub>2</sub>O), 347 (M\*-HCO), 290 (M\*-C<sub>4</sub>H<sub>6</sub>O<sub>2</sub>), 273, 261, 259, 243 and 69 (C<sub>4</sub>H<sub>5</sub>O). (Calc. for C<sub>20</sub>H<sub>24</sub>O<sub>7</sub>: C, 63.82; H, 6.43. Found: C, 63.56; H, 6.18%).

The most polar substance was identified as darutigenol (8a), yield 0.48 g, mp  $168^\circ$ ,  $[\alpha]_D-12^\circ$  (c, 1.024), reported [2] mp  $168-170^\circ$ ,  $[\alpha]_D-11^\circ$ , MS m/e: 322, 305, 304, 291, 286, 273, 271, 261, 227, 187, 173, 135, 120, 109, 107, 105. (Calc. for  $C_{20}H_{34}O_3$ : C, 74.49; H, 10.63. Found: C, 74.32; H, 10.41%). Gummy triacetate 8b, MS m/e 448, tribenzoate 8c mp 82°, reported [2, 3] mp 83°. Degradation of 8a (0.15 g with 0.1 g sodium periodate in MeOH for 12 hr and recrystallization of the crude product from MeOH gave 0.102 g of 9, mp 118–120°, reported [2, 3] mp 115–120°, MW (MS) 290

Fractions 25–36 exhibited a single spot on TLC, and were combined and recrystallized from EtOH to give 2.6 g of darutoside (8d), mp 250°,  $[\alpha]_D$  – 35°, reported [2, 3] mp 248–250°,  $[\alpha]_D$  – 37°, hexaacetate mp 92–94°, reported [2] mp 91–93°. Oxidation of 0.25 g of 8d in MeOH with 0.20 g of NaIO<sub>4</sub> for 12 hr followed by hydrolysis of the crude product with KOH in EtOH and acidification also gave 9, mp 118–120°.

Reactions of 1b. A mixture of 0.10 g 1b, 10 ml MeOH and 0.5 ml 40% KOH was stirred for 3 hr under N<sub>2</sub>, acidified with HOAc and extracted with CHCl<sub>3</sub>. The washed and dried extract was evapd; the residue was purified by PLC (C<sub>6</sub>H<sub>6</sub>-EtOAc, 1:2) and recrystallized from EtOAc. Yield of 2a 40 mg, mp 135-140°, IR  $\nu_{\rm max}$  cm<sup>-1</sup>: 3500, 1770, 1690 and 1100; UV  $\lambda_{\rm max}^{\rm MeOH}$  nm: 230 (ε 9600), <sup>1</sup>H NMR (60 MHz): 89.50 (d, J = 2 Hz, H-14), 6.80 (dd, J = 9, 8 Hz, H-1), 5.0 (m, H-5 and H-6) 3.5-3.95 (H-8, H-9, H-13), 3.40 and 3.25 (OMe), 1.98 (br, H-15); MS m/e: 324 (M<sup>+</sup>), 306, 295, 292, 277, 274, 245, 242, 213. (Calc. for C<sub>17</sub>H<sub>24</sub>O<sub>6</sub>: C, 62.95; H, 7.46. Found: C, 62.81; H, 7.24%).

Acetylation (Ac<sub>2</sub>O-Py) of 20 mg **2a** gave 20 mg **2b**, mp 170° (from MeOH). IR bands at 1770, 1720, 1690 and 1160 cm<sup>-1</sup>; <sup>1</sup>H NMR:  $\delta$  9.45 (d, J = 2 Hz, H-14), 6.70 (dd, J = 9, 8 Hz, H-1), 6.20 (dbr, J = 9, 2 Hz, H-8), 5(m, H-5 and H-6), 3.6-3.8 (H-9 and H-13), 3.35 and 3.15 (OMe), 2.15 (Ac) and 1.96 (dbr, H-15); MS m/e: 366 (M<sup>+</sup>), 337, 334, 324, 306, 277, 274. (Calc. for C<sub>19</sub>H<sub>26</sub>O<sub>7</sub>: C, 62.28; H, 7.15. Found: C, 62.12; H, 6.95%).

Oxidation of 2a. A soln of 30 mg 2a in 1 ml DMSO and 1 ml Ac<sub>2</sub>O was kept overnight at room temp, diluted with  $H_2O$  and extracted with CHCl<sub>3</sub>. The washed and dried extract was evapd and the residue purified by PLC to give 20 mg 3, mp 195-200° after 4 recrystallizations from MeOH. IR  $\nu_{\text{max}}$  cm<sup>-1</sup>: 2800, 1780, 1720 and 1690; UV strong end absorption ( $\varepsilon_{230}$  12 000); <sup>1</sup>H NMR (270 MHz):  $\delta$  9.50 (H-14), 7.01 (ddbr, J = 9, 8 Hz, H-1), 5.16 (d, J = 10 Hz, H-5),

4.64 (t, J = 10 Hz, H-6), 3.87 (dd, 9.5, 1 Hz) and 3.44 (dd, J = 9.5, 2 Hz, H-13), 3.87 (H-9), 3.37 and 3.33 (OMe) superimposed on two proton multiplet of H-7 and H-11, 2.72 (m, H-2a), 2.49 (m, H-2b and H-3a), 2.22 (t, J = 13 Hz, H-3b), 1.90 (br, H-15); MS m/e: 322 ( $M^+$ ), 294, 262, 251, 249, 231. (Calc. for  $C_{17}H_{22}O_6$ : MW, 322.1415. Found: MW(MS), 322.1414).

Reactions of 4a. Acetylation of 30 mg 4a with  $Ac_2O-Py$  gave 30 mg 4b as a gum which had IR bands at 2800, 1770, 1720, 1690 a d 1180 cm<sup>-1</sup>;  $^1H$  NMR (60 MHz):  $\delta$  9.50 (d, J=2 Hz, H-14), 6.80 (dd, J=9, 8 Hz, H-1), 6.60 (dd, J=9, 1 Hz, H-8), 6.26 and 5.88 (d, J=3 Hz, H-13), 6.10 and 5.60 (br, H-3'), 5.20 (m, H-5 and H-6), 4.85 (br, H-15), 3.80 (dd, J=9, 2 Hz, H-9), 3.15 (OMe), 2.10 (Ac) and 1.98 (br, H-4'); MS m/e: 418 (M<sup>+</sup>), 389, 387, 353, 349, 333, 289, 272, 243, 211 and 69.

A soln of 0.1 g 4a in 10 ml MeOH was hydrolysed with 40% aq. KOH as described for 1b. The product (2c) was recrystallized from ETOAc and melted at 90-92°, mmp undepressed with material fron hydrolysis of 1a [1]; 1H NMR and MS identical. Oxidation of 25 mg 2c with MnO<sub>2</sub> as previously described [1] gave dialdehyde 6a as a gum whose <sup>1</sup>H NMR spectrum was not described earlier; signals appeared at  $\delta$  10.17 (br, H-15), 9.47 (d, J = 2 Hz, H-14), 6.85 (dd, J = 9, 8 Hz, H-1), 5.90 d (br, J = 9 Hz, H-5), 4.85 (dbr,J = 9 Hz, H-6), 3.38 and 3.15 (OMe). Oxidation of 20 mg 6a with DMSO-Ac<sub>2</sub>O as described for 2a gave 15 mg 6b as a gum which had IR bands at 2820, 2620, 1780, 1720, 1680, 1625, 1100, 1050 and 1000 cm<sup>-1</sup>; <sup>1</sup>H NMR; δ 10.23 (H-15), 9.66 (H-14), 7.10 (dd, J = 9, 8 Hz, H-1), 6.40 (d, J = 10 Hz, H-5), 5.45 (dd, J = 10, 9 Hz, H-6), 3.70 (H-9), 3.40 and 3.35 (OMe); MS m/e: 336 (M<sup>+</sup>), 308, 276, 263 and 245. (Calc. for  $C_{17}H_{20}O_7$ : MW, 336.1208. Found: MW (MS), 336.1204).

A mixture of 0.04 g **4b**, 2 ml  $Ac_2O$  and 5.5 ml  $BF_3$ -etherate was left overnight at room temp. poured into cold  $NaHCO_3$  soln and extracted with EtOAc. The washed and dried extract was evapd and the residue purified by PLC ( $C_6H_6$ -EtOAc, 1:1) to give 25 mg as a gum which had IR bands at 1770, 1720 and 1160 cm<sup>-1</sup>;  $^1H$  NMR (60 MHz):  $\delta$  7.22 (H-14), 6.38 and 5.85 (d, J=3 Hz, H-13), 6.1 (overlapping signals of H-1, H-8 and H-3'), 5.65 (br, H-3'), 5.28 (m, H-5 and H-6), 4.90 (H-15), 3.90 (d, J=9 Hz, H-9), 3.15 (OMe), 2.20, 2.15, 2.15 (Ac), 2.0 (br, H-4'); MS m/e: 520 ( $m^+$ ), 489, 461, 434, 387, 359, 349, 333, 289, 273, 272, 241, 213, 212, 211 and 69.

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